Received 12 October 2005

Accepted 17 October 2005

Online 31 October 2005

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

Ning-Ning Pan,^a Wei Wang^a* and Bing Zhao^b

^aSchool of Chemical Engineering, Anshan University of Science and Technology, Anshan 114002, People's Republic of China, and ^bSchool of Chemical Engineering and Technology, Tianiin University, Tianiin 300072. People's Republic of China

Correspondence e-mail: panningning@163.com

Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.004 Å R factor = 0.040 wR factor = 0.122 Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The two terminal 1-methylimidazol-2-vlsulfanyl groups adopt a syn conformation with respect to the central quinoxaline ring system in the title compound, $C_{18}H_{18}N_6S_2$.

quinoxaline

Comment

To date, a large number of flexible or rigid chain-linked dithioether ligands containing N-heterocyclic groups have been synthesized and investigated, due to their diverse coordination capabilities and to the important properties of their metal complexes (Hester et al., 1997; Yang et al., 2000; Bu et al., 2002; Hong et al., 2000; Tong et al., 1998). In this context, a new imidazole derivative, viz. 2,3-bis(1-methyl-1H-imidazol-2ylsulfanylmethyl)quinoxaline, (I), has been synthesized.

2,3-Bis(1-methyl-1H-imidazol-2-ylsulfanylmethyl)-



In the molecular structure of (I) (Fig. 1 and Table 1), the two terminal 1-methylimidazol-2-ylsulfanyl groups adopt a syn conformation with respect to the central quinoxaline ring system and the dihedral angle formed between the imidazole rings is $128.7 (3)^{\circ}$. The central quinoxaline ring system is practically planar, forming dihedral angles of 130.9 (3) and $27.7 (3)^{\circ}$ with the imidazoles attached to S2 and S1, respectively, with the imidazole rings. In one imidazole ring, atom C10 attached to the sulfanyl group has a distorted trigonal geometry, with the N3-C10-N4 [112.0 (2) $^{\circ}$] and N4-C10-S1 [126.74 (19)°] angles deviating significantly from the ideal sp^2 -hybridized values; the comparable values for the other imidazole ring are 111.8 (2) $^{\circ}$ for N5-C15-N6 and 125.0 (2) $^{\circ}$ for N6-C15-S2. Owing to $p-\pi$ conjugation, the Csp²-S bonds, *i.e.* S1-C10 and S2-C15, are significantly shorter, as expected, than the Csp³-S bonds, *i.e.* S1-C9 and S2-C14 (see Table 1). The average lengths for the Csp^2 -S and Csp^3 -S bonds are 1.745 (3) and 1.814 (3) Å, respectively, which are comparable to those reported in the literature (Zhang et al., 2003; Zheng & Liu, 2003).

© 2005 International Union of Crystallography Printed in Great Britain - all rights reserved

Acta Cryst. (2005). E61, o3939-o3940

organic papers

Experimental

A solution of 2,3-bis(bromomethyl)quinoxaline (1.58 g, 5 mmol) in ethanol (10 ml) was added dropwise to a mixture of 1-methyl-2thioimidazole (1.26 g, 11 mmol), KOH (0.62 g, 11 mmol) and ethanol (10 ml). The reaction mixture was stirred for 24 h at room temperature. Water (30 ml) was gradually added after which a yellow precipitate appeared. This was filtered off and recrystallized from an ethanol and water (1:1 v/v) mixture (yield 70%, m.p. 374–375 K). Analysis found: C 56.48, H 4.69, N 22.05%; C₁₈H₁₈N₆S₂ requires C 56.52, H 4.74, N 21.97%. IR (KBr): v 3103, 2940, 1484, 1460, 1410, 1372, 1272, 1118, 1014, 760, 676 cm^{-1. 1}H NMR (CDCl₃): δ 3.45 (s, 3H), 4.61 (s, 2H), 6.86 (d, 1H, J = 1.5 Hz), 7.00 (d, 1H, J = 2.5 Hz), 7.70 (d, 1H, J = 4 Hz), 7.91 (d, 1H, J = 3.5 Hz). Crystals of (I) suitable for single-crystal X-ray analysis were obtained by slow evaporation of a mixture of ethanol and water (1:2 v/v).

Z = 2

 $D_x = 1.383 \text{ Mg m}^{-3}$

Cell parameters from 1798

 $0.28 \times 0.24 \times 0.14$ mm

 $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\theta=2.9{-}26.1^\circ$ $\mu = 0.31 \text{ mm}^{-1}$

T = 294 (2) K

Block, yellow

Crystal data

C18H18N6S2 $M_r = 382.50$ Triclinic, P1 a = 6.8978 (14) Å b = 9.953 (2) Å c = 14.556 (3) Å $\alpha = 96.495 (4)^{\circ}$ $\beta = 97.418 \ (3)^{\circ}$ $\gamma = 109.863 (3)^{\circ}$ V = 918.7 (3) Å³

Data collection

3244 independent reflections
2350 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.021$
$\theta_{\rm max} = 25.0^{\circ}$
$h = -7 \rightarrow 8$
$k = -9 \rightarrow 11$
$l = -17 \rightarrow 16$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.067P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	+ 0.146P]
$wR(F^2) = 0.122$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} = 0.002$
3244 reflections	$\Delta \rho_{\rm max} = 0.24 \text{ e } \text{\AA}^{-3}$
237 parameters	$\Delta \rho_{\rm min} = -0.24 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

S1-C9	1.803 (3)	S2-C14	1.825 (3)
S1-C10	1.754 (2)	S2-C15	1.736 (3)
S1 C10 N2	121 20 (19)	50 C15 N5	102 12 (10)
SI-CIU-N3	121.20 (18)	52-C15-N5	123.13 (19)
S1-C10-N4	126.74 (19)	S2-C15-N6	125.0 (2)
N3-C10-N4	112.0 (2)	N5-C15-N6	111.8 (2)





H atoms were included in the riding-model approximation, with aromatic C-H = 0.93 Å, methylene C-H = 0.97 Å and methyl C-H = 0.96 Å, and with $U_{iso}(H) = 1.2U_{eq}(aromatic and methylene C)$ and $U_{iso}(H) = 1.5U_{eq}$ (methyl C).

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

References

- Bruker (1997). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bu, X. H., Chen, W., Du, M., Kumar, B., Wang, W. Z. & Zhang, R. H. (2002). Inorg. Chem. 41, 437-439.
- Hester, C. A., Baughman, R. G. & Collier, H. L. (1997). Polyhedron, 16, 2893-2895
- Hong, M. C., Zhao, Y. J., Su, W. P., Cao, R., Fujita, M., Zhou, Z. Y. & Chan, A. S. C. (2000). Angew. Chem. Int. Ed. 39, 2468-2470.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany,
- Tong, M. L., Ye, B. H., Cai, J. W., Chen, X. M. & Ng, S. W. (1998). Inorg. Chem. 37, 2645-2650.
- Yang, S. P., Zhu, H. L., Yin, X. H., Chen, X. M. & Ji, L. N. (2000). Polyhedron, 19. 2237-2242.
- Zhang, W., Liu, H.-M., Li, C.-B. & Zhang, W.-Q. (2003). Acta Cryst. E59, o26-027
- Zheng, Y. & Liu, H.-B. (2003). Acta Cryst. E59, 034-035.